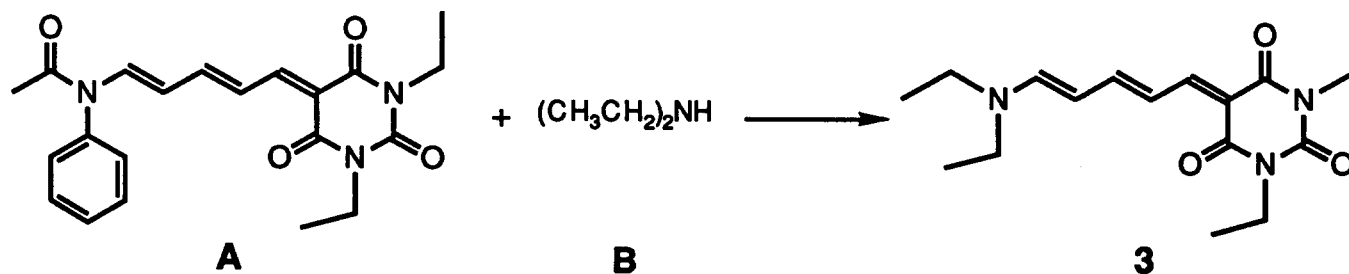


J-2620-m1

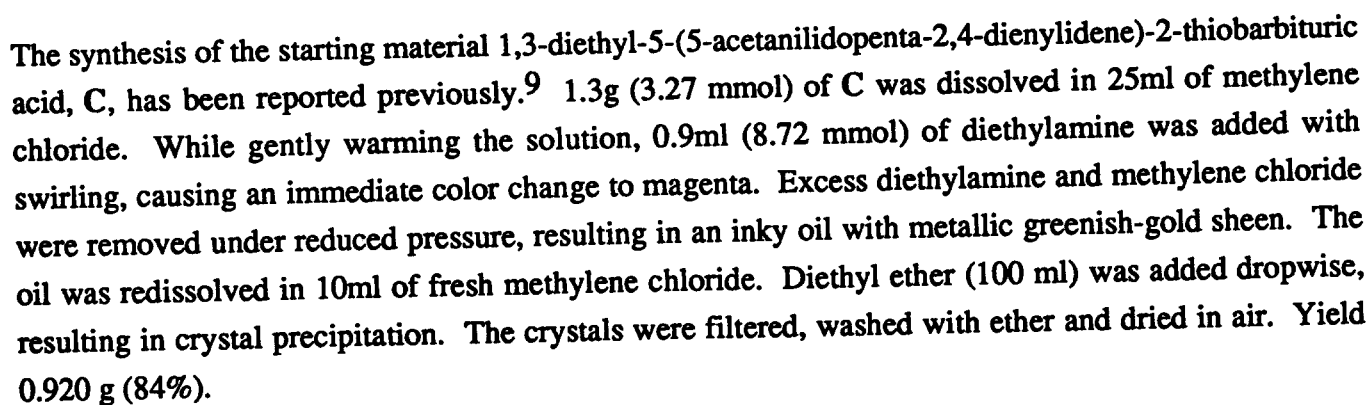
Synthesis of 1,3-diethyl-5-(5-(N,N'-diethylamino)penta-2,4-dienylidene) barbituric acid, **3**.



The synthesis of the starting material 1,3-diethyl-5-(5-acetanilidopenta-2,4-dienylidene) barbituric acid, **A**, has been reported previously.<sup>9</sup> 1.3g (3.41 mmol) of **A** was dissolved in 25ml of methylene chloride. While gently warming the solution, 0.9ml (8.72 mmol) of diethylamine was added with swirling, causing an immediate color change to magenta. Excess diethylamine and methylene chloride were removed under reduced pressure, resulting in an inky oil with a metallic greenish sheen. The oil was redissolved in 10ml of fresh methylene chloride. Diethyl ether (100 ml) was added dropwise, resulting in crystal precipitation. The crystals were filtered, washed with ether and dried in air. Yield 0.950 g (87%).

Characterizing data for **3**: <sup>1</sup>H NMR.  $\delta$  8.00 (d,  $J = 13.3$  Hz, 1H), 7.72 (dd,  $J = 13.3, 13.2$  Hz, 1H), 7.26 (dd,  $J = 13.2, 12.2$  Hz, 1H), 7.05 (d,  $J = 12.3$  Hz, 1H), 5.68 (dd,  $J = 12.3, 12.2$  Hz, 1H), 4.00 (q,  $J = 7.0$  Hz, 4H), 3.37 (m, 4H), 1.27 (m, 6H), 1.23 (t,  $J = 7.0$  Hz, 3H), 1.22 (t,  $J = 7.0$  Hz, 3H). Elemental Analysis: Calculated for C<sub>17</sub>H<sub>25</sub>N<sub>3</sub>O<sub>3</sub>: C, 63.93; H, 7.89; N, 13.16; Found: C, 63.92; H, 7.91; N, 13.19.

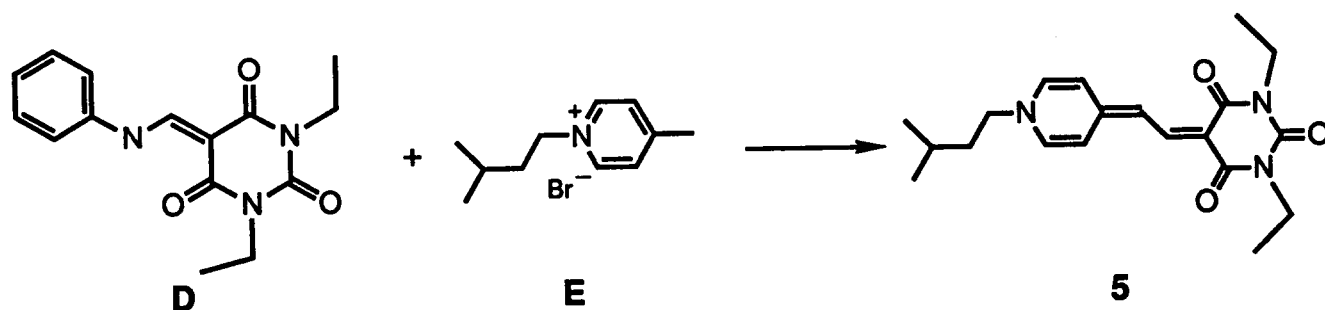
### Synthesis of 1,3-diethyl-5-(5-(N,N'-diethylamino)penta-2,4-dienylidene)-2-thiobarbituric acid, 4.



**Characterizing data for 4:**  $^1\text{H}$  NMR.  $\delta$  7.99 (d,  $J = 13.5$  Hz, 1H), 7.79 (dd,  $J = 13.5, 12.3$  Hz, 1H), 7.33 (dd,  $J = 12.4, 13.0$  Hz, 1H), 7.18 (d,  $J = 12.2$  Hz, 1H), 5.79 (dd,  $J = 12.2, 12.4$  Hz, 1H), 4.57 (q,  $J = 6.9$  Hz, 4H), 3.42 (q,  $J = 7.1$  Hz, 4H), 1.30 (m, 12H). Elemental Analysis: Calculated for  $\text{C}_{17}\text{H}_{25}\text{N}_3\text{O}_2\text{S}$ : C, 60.87; H, 7.51; N, 12.53; S, 9.56. Found: C, 60.77; H, 7.56; N, 12.44; S, 9.63.

J-2620-m3

Synthesis of 1,3-diethyl-5-[(1-isopentyl-4-pyridylidene)-ethylene]-barbituric acid, **5**.

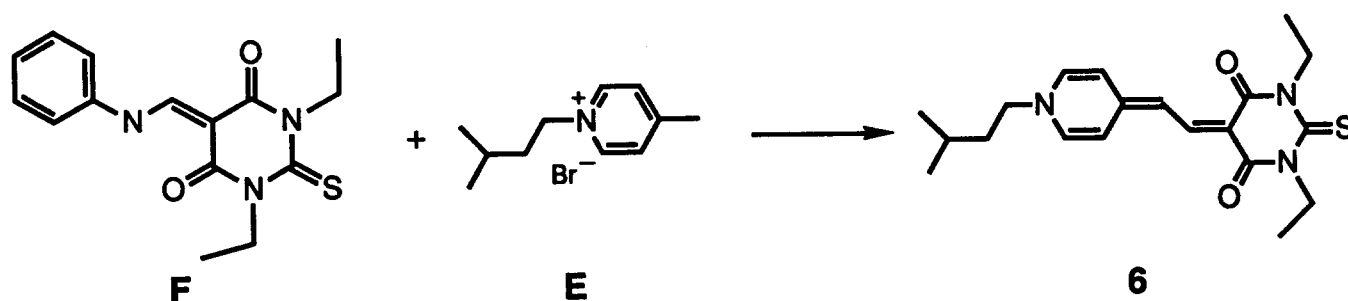


The synthesis of the starting material 1,3-diethyl-5-anilinyldene barbituric acid, **D**, has been reported previously.<sup>9</sup> 3.97g (16.2 mmol) of isopentyl-4-picolinium bromide, **E**, and 4.9g (17.1 mmol) of **D** were added to 50ml toluene while gently heating the solution. Then, 2.5ml of triethylamine was added and the reaction mixture refluxed for 2.5 hours at 110°C. Toluene was removed under vacuum and water was added to the remaining residue. An orange precipitate formed upon addition of the water. This precipitate was dissolved in toluene and dichloromethane and separated from the water layer. The solution was dried over Magnesium Sulfate, filtered and the remaining solvent removed under vacuum. The residue was recrystallized from toluene. Yield 1.98g (34.2%).

Characterizing data for **5**: <sup>1</sup>H NMR.  $\delta$  8.38 (d,  $J = 15.0$  Hz, 1H), 7.36 (d,  $J = 14.8$  Hz, 1H), 7.36 (d,  $J = 7.2$  Hz, 2H), 7.6 - 7.0 (very broad peak, due to hindered rotation, 2H), 4.00 (q,  $J = 7.0$  Hz, 4H), 3.90 (m, 2H), 1.72 (m, 2H), 1.67 (m, 1H), 1.22 (t,  $J = 6.9$  Hz, 6H), 0.98 (d,  $J = 6.6$ , 6H). Elemental Analysis: Calculated. for C<sub>20</sub>H<sub>27</sub>N<sub>3</sub>O<sub>3</sub>: C, 67.20; H, 7.61; N, 11.76; Found: C, 67.32; H, 7.60; N, 11.76.

J-2620-m4

Synthesis of 1,3-diethyl-5-[(1-isopentyl-4-pyridylidene)-ethylene]-2-thiobarbituric acid, **6**.

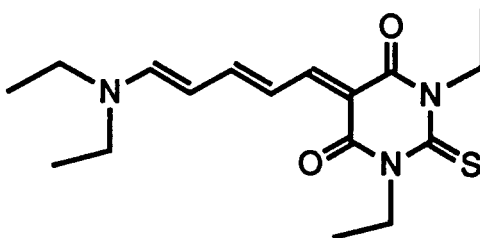


The synthesis of the starting material 1,3-diethyl-5-anilinyldene-2-thiobarbituric acid, **F**, has been reported previously.<sup>9</sup> 3.60g (14.8 mmol) of isopentyl-4-picolinium bromide, **E**, and 4.48g (14.8 mmol) of **F** were added to 50ml toluene while gently heating the solution. Then, 2.5ml of triethylamine was added and the reaction mixture refluxed for 2 hours at 110°C. Toluene was removed under vacuum and water was added to the remaining residue. A red precipitate formed upon addition of the water. This precipitate was dissolved in toluene and dichloromethane and separated from the water layer. The solution was dried over Magnesium Sulfate, filtered and the remaining solvent removed under vacuum. The residue was recrystallized from toluene to give red crystals. Yield 2.43g (44%).

Characterizing data for **6**: <sup>1</sup>H NMR. δ 8.36 (d, J = 15.3 Hz, 1H), 7.64 (d, J = 15.1 Hz, 1H), 7.49 (d, J = 7.3 Hz, 2H), 7.34 (broad peak, due to hindered rotation, 2H), 4.60 (q, J = 6.9 Hz, 4H), 3.98 (m, 2H), 1.75 (m, 2H), 1.63 (m, 1H), 1.31 (t, J = 6.9 Hz, 6H), 0.99 (d, J = 6.7, 6H). Elemental Analysis: Calculated. for C<sub>20</sub>H<sub>27</sub>N<sub>3</sub>O<sub>2</sub>S: C, 64.31; H, 7.29; N, 11.25; S, 8.58; Found: C, 64.21; H, 7.26; N, 11.28; S, 8.50.

J-2620-m5

X-RAY CRYSTALLOGRAPHIC SUPPLEMENTARY MATERIAL FOR  
COMPOUND 4



1,3-diethyl-5-(5-(N,N'-diethylamino)penta-2,4-dienylidene)-2-thiobarbituric acid

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Name	1,3-diethyl-5-(5-(N,N'-diethylamino)penta-2,4-dienylidene)-2-thiobarbituric acid
Formula	C <sub>17</sub> H <sub>25</sub> N <sub>3</sub> O <sub>2</sub> S
Formula Weight	335.46
Crystal System	triclinic
Space Group	P $\bar{1}$
Cell Dimensions a, Å	10.272 (2)
b, Å	10.288 (1)
c, Å	10.291 (2)
$\alpha$ , °	60.97 (1)
$\beta$ , °	87.03 (1)
$\gamma$ , °	70.57 (1)
V, Å <sup>3</sup>	888.8 (3)
Z	2
Density, calc, g cm <sup>-3</sup>	1.26
Crystal color, habit	red, irregular prism
Crystal size, mm <sup>3</sup>	0.33 x 0.35 x 0.56
$\mu$ , cm <sup>-1</sup>	1.86
$\mu$ r <sub>max</sub>	0.71073
maximum 2 $\theta$ (scan type)	28°
range of h,k,l	-13-13, -13-13, -13-13
# of reflections measured	8687
# of independent reflections	4292
# reflections, $F_o^2 > 0$	3973
# reflections, $F_o^2 > 3\sigma(F_o^2)$	3326
GOF, merge	1.01
R(merge) for refs meas. twice	0.019
secondary extinction (x 10 <sup>-6</sup> )	2.16(29)
R, $F_o^2 > 0$	0.046
R, $F_o^2 > 3\sigma(F_o^2)$	0.038
GOF (number of parameters)	2.34, (309)
( $\Delta/\sigma$ ) <sub>max</sub> in final least squares	<0.005
Final Difference map:	
Maximum, eÅ <sup>-3</sup>	+ 0.34
Minimum, eÅ <sup>-3</sup>	- 0.28

Data were collected at 296 K.

Hydrogen positions were assumed, C-H 0.95 Å, and repositioned once near the end of refinement.

Structure solved using MULTAN 88.

J-2620-m7

X-ray I.D. SRM31

Name of Compound Diethylamino-II-Thiobarbituric acid

Chemical Formula C<sub>17</sub>H<sub>25</sub>N<sub>3</sub>O<sub>2</sub>S M. wt. 335.46

Crystal System Triclinic Space Group P $\bar{1}$  (# 2)

a= 10.272(2) Å  $\alpha$ = 60.97(1)°

b= 10.288(1) Å  $\beta$ = 87.03(1)°

c= 10.291(2) Å  $\gamma$ = 70.57(1)°

V= 888.8(3) Å<sup>3</sup> Z= 2

D<sub>m</sub> — g cm<sup>-3</sup> D<sub>x</sub> 1.26 g cm<sup>-3</sup>

Radiation used MoK $\alpha$  Wavelength 0.71073 Å

Absorption Coefficient,  $\mu$ = 1.86 cm<sup>-1</sup> Temperature 296 °K

Type of Absorption Correction none (program used) —

Range of Transmission Factors —

Crystal Color red Crystal Shape (Habit) irregular prism

Crystal Size 0.33 mm  $\times$  0.35 mm  $\times$  0.56 mm

Source of Crystal synthesized by BGT

Type of Diffractometer Enraf-Nonius Cad-4

Data Collection method (diffraction geometry, scan type)  $\theta$ -2 $\theta$  scans

Lattice Parameters: Number of reflections 25;  $\theta$  range 4° to 16°

$\theta$  range for data collection 1° to 28°

$h_{min}$  -13  $h_{max}$  13  $k_{min}$  -13  $k_{max}$  13  $l_{min}$  -13  $l_{max}$  13

Number of reflections measured 8687

Number of independent reflections 4292

Number of reflections used in refinement 4292

Criterion for reflections used all used,  $F_o^2$  positive and negative

Goodness of fit for merging data 1.01 (number of multiples 4292)

J-2620-m8

$R_{int}$  for duplicate reflections 0.019 (number of duplicates 3815)

Number of standard reflections 3 Interval 150 minutes

Variations of standards within counting statistics

How structure solved? Sharpened Patterson map gave vectors which were interpreted to give positions of ring atoms and the sin connected to them; remaining atoms found from subsequent Structure Factor - Fourier calculation.

How H atoms treated? all refined with isotropic displacement parameters

Refinement on  $F^2$ ,  $w=1/\sigma^2(F_o^2)$ , one full matrix used.

$R = 0.046$  on  $F$  for 3973 reflections with  $F_o^2 > 0$

$R = 0.038$  on  $F$  for 3326 reflections with  $F_o^2 > 3\sigma(F_o^2)$

$wR = 0.008$  on  $F^2$  for 4292 reflections

Goodness of Fit ( $S$ ) = 2.34 for 4292 data and 309 parameters

$(\Delta/\sigma)_{max}$  in final least squares cycle < 0.005

$\Delta\rho_{max}$  +0.34  $e\text{\AA}^{-3}$ ,  $\Delta\rho_{min}$  -0.28  $e\text{\AA}^{-3}$  in final difference map.

Secondary Extinction parameter (if used)  $2.16(29) \times 10^{-6}$

(Reference: Larson, A. C. (1967). *Acta Cryst.* 23, 644-665.)

### Computer Programs

The CRYM Crystallographic Computing System

(Duchamp, D. J. (1964). *Am. Crystallogr. Assoc. Meet.*, Bozeman, Montana, Paper B14, p. 29.)

### ORTEP

(Johnson, C. K. (1976). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Oak Ridge, Tennessee, USA.)



## MULTAN88

Debaerdemaeker, T., Germain, G., Main, P., Refaat, L. S., Tate, C. & Woolfson, M. M. (1988). *MULTAN 88. Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*, Univs. of York, England and Louvain, Belgium.

Scattering Factors and  $f'$ ,  $f''$ :

Cromer, D. T. (1974). *International Tables For X-ray Crystallography*, Vol. IV, pp. 149-151. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)

Cromer, D. T. & Waber, J. T. (1974). *International Tables For X-ray Crystallography*, Vol. IV, pp. 99-101. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)

## Any additional Data:

*Structure solved and refined  
by R. E. Marsh.*

Weights  $w$  are calculated as  $1/\sigma^2(F_o^2)$ ; variances ( $\sigma^2(F_o^2)$ ) were derived from counting statistics plus an additional term,  $(0.014I)^2$ ; variances of the merged data were obtained by propagation of error plus another additional term,  $(0.014\bar{I})^2$ .

## Definitions:

$$R = \frac{\sum |F_o - |F_c||}{\sum F_o}; \quad R_w = \frac{\sum w(F_o^2 - F_c^2)^2}{\sum w(F_o^2)^2}$$

$$S = \left\{ \frac{\sum w(F_o^2 - F_c^2)^2}{n - p} \right\}^{\frac{1}{2}}$$

where  $n$  = number of data,

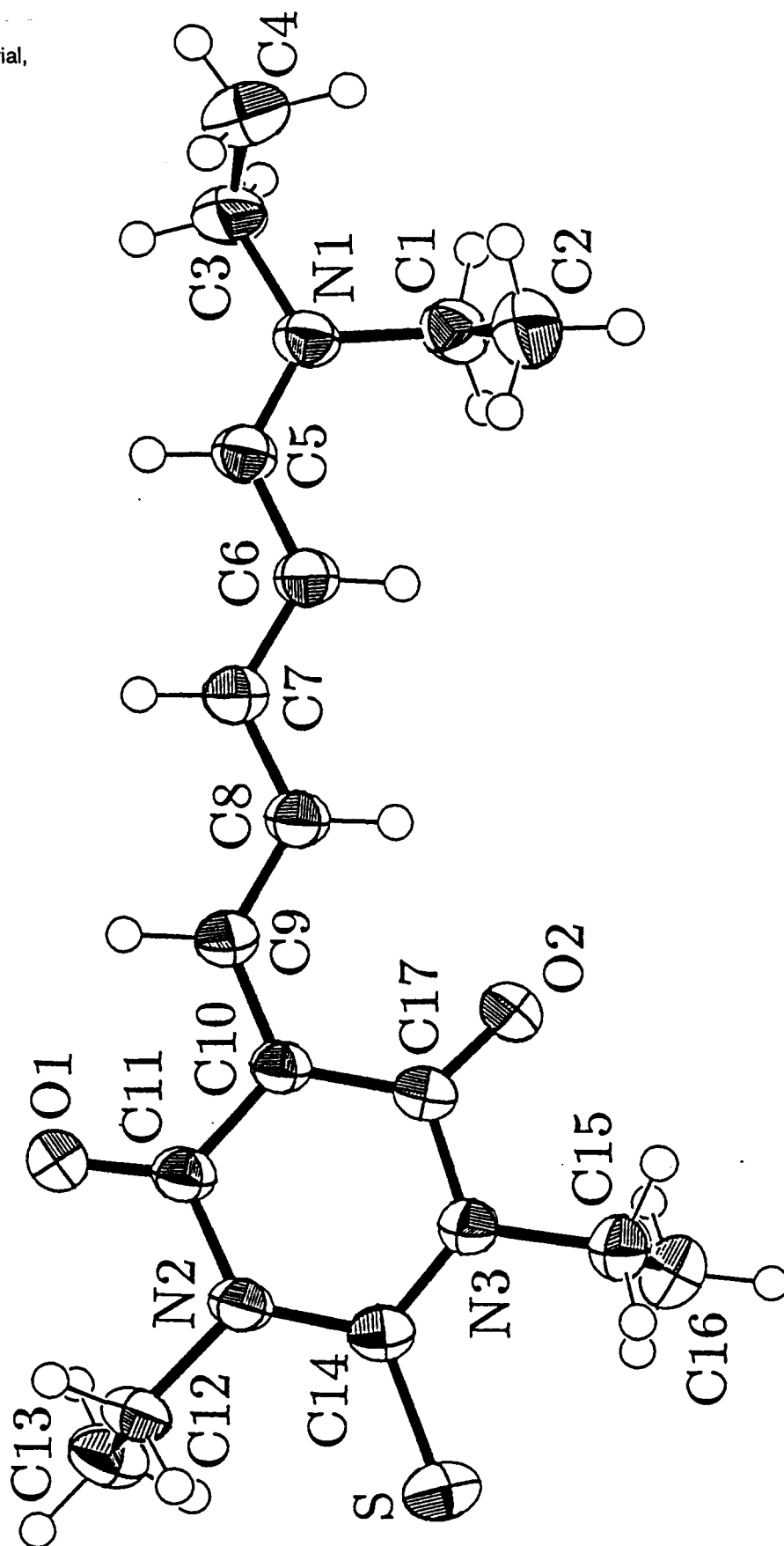
$p$  = number of parameters refined.

## Legends for Figures.

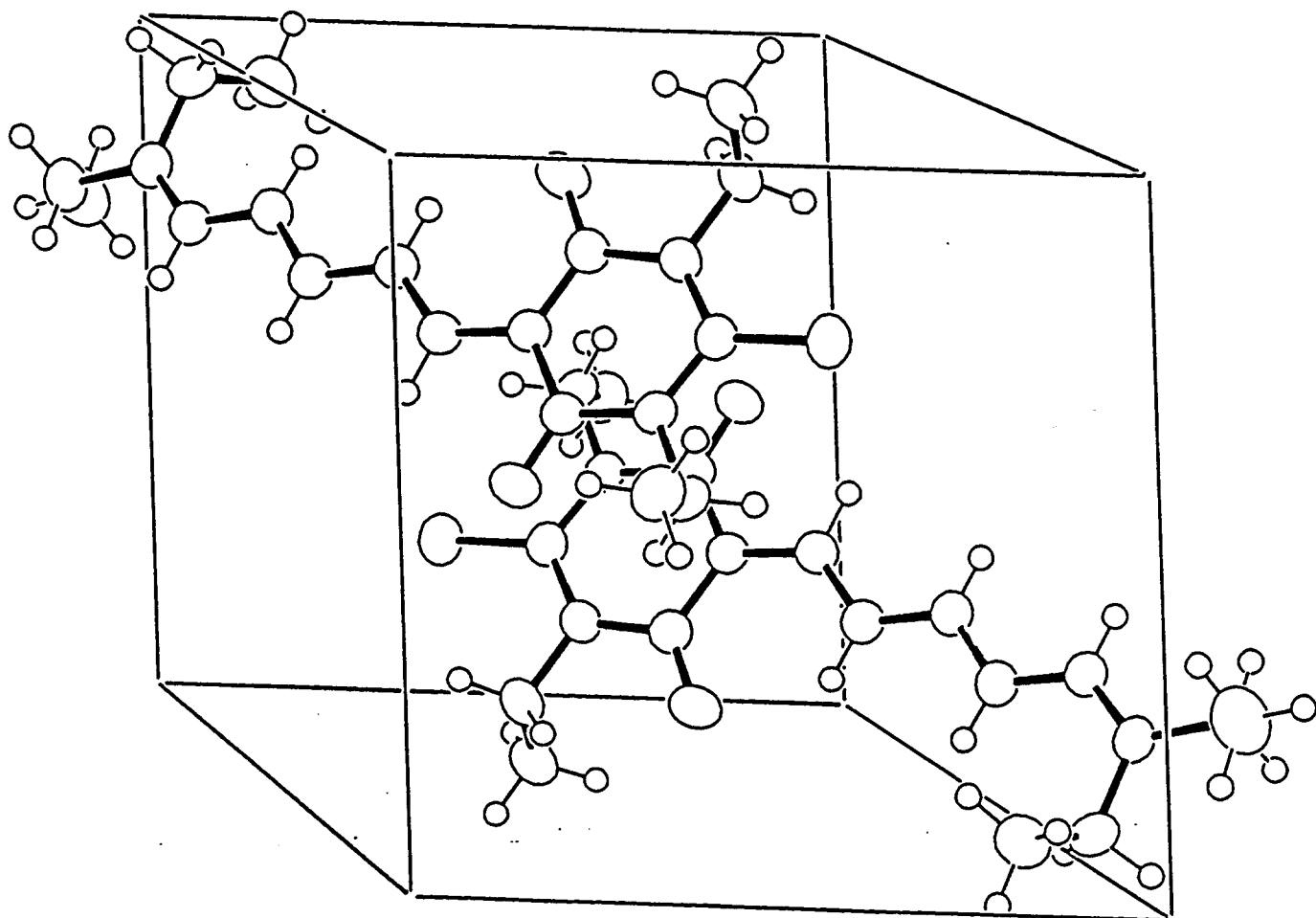
Figure 1. An ORTEP drawing of the molecule with 50 % probability ellipsoids showing the numbering system. Hydrogen atoms are shown with arbitrary, small displacement parameters.

Figure 2. An ORTEP drawing of the contents of a unit cell with a unit cell outlined. Atoms are shown as 50 % probability ellipsoids; hydrogen atoms are as in Fig. 1. The view is perpendicular to the a c plane.

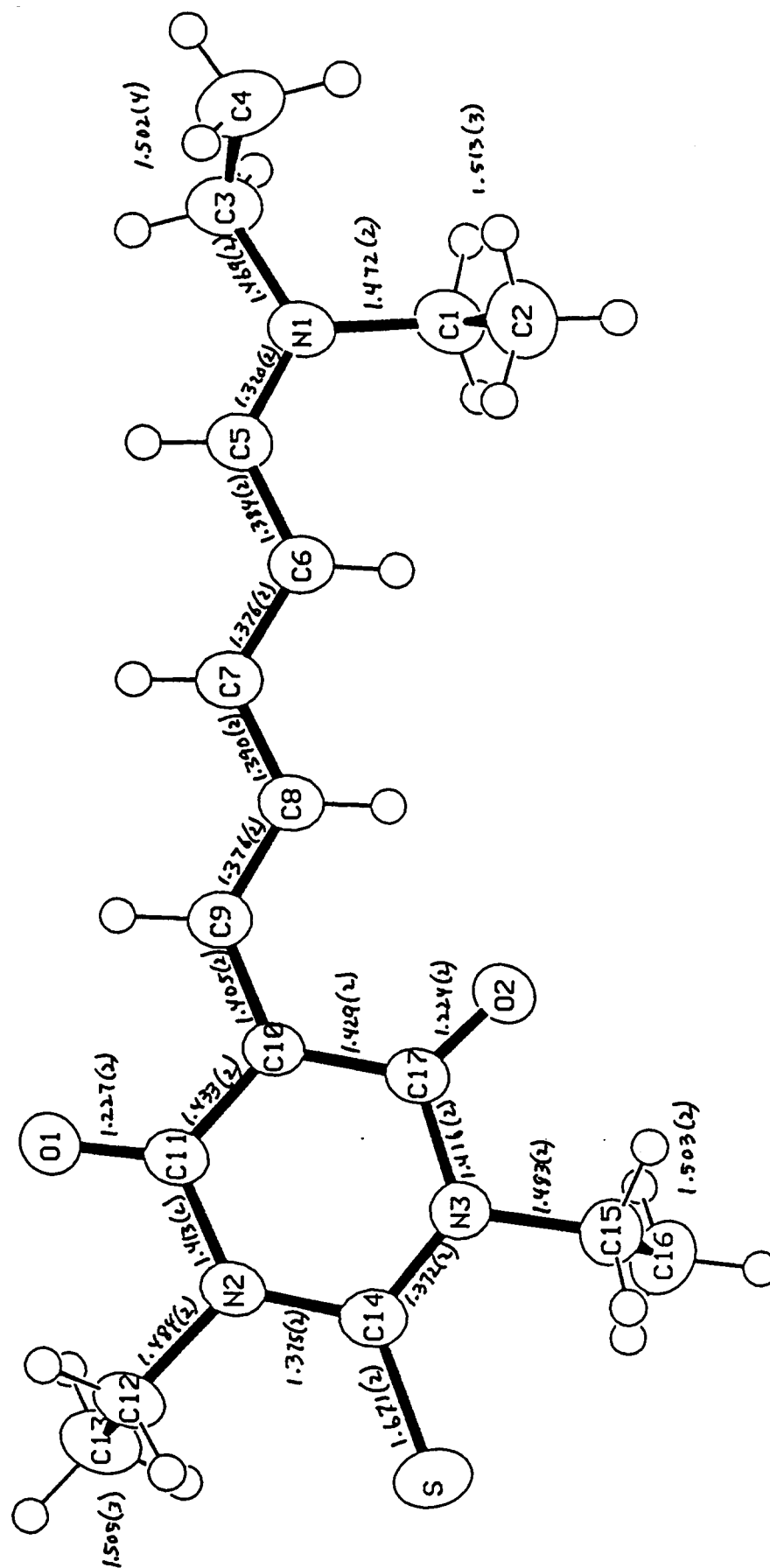
J-2620-m11



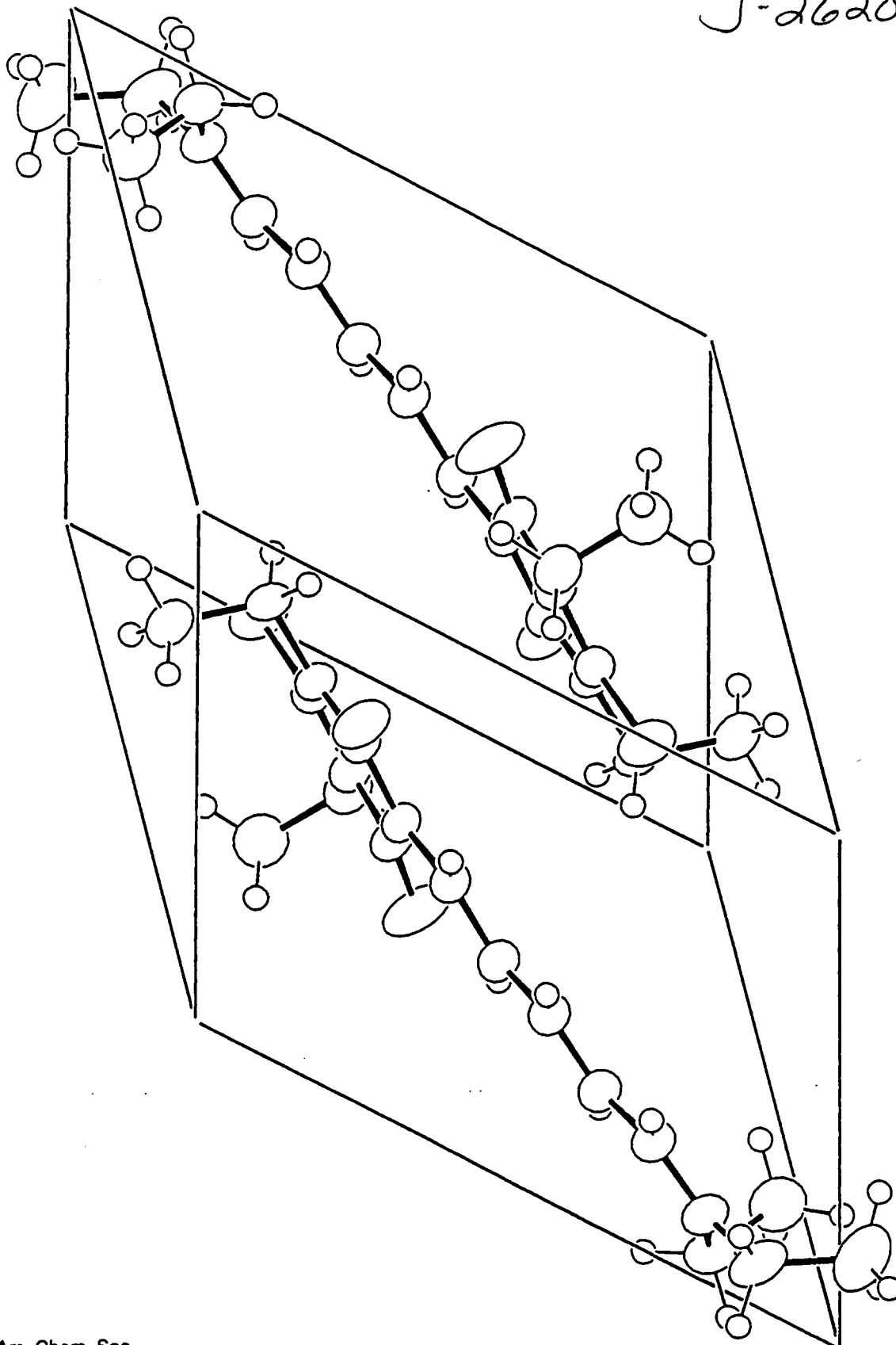
J-2620-m12



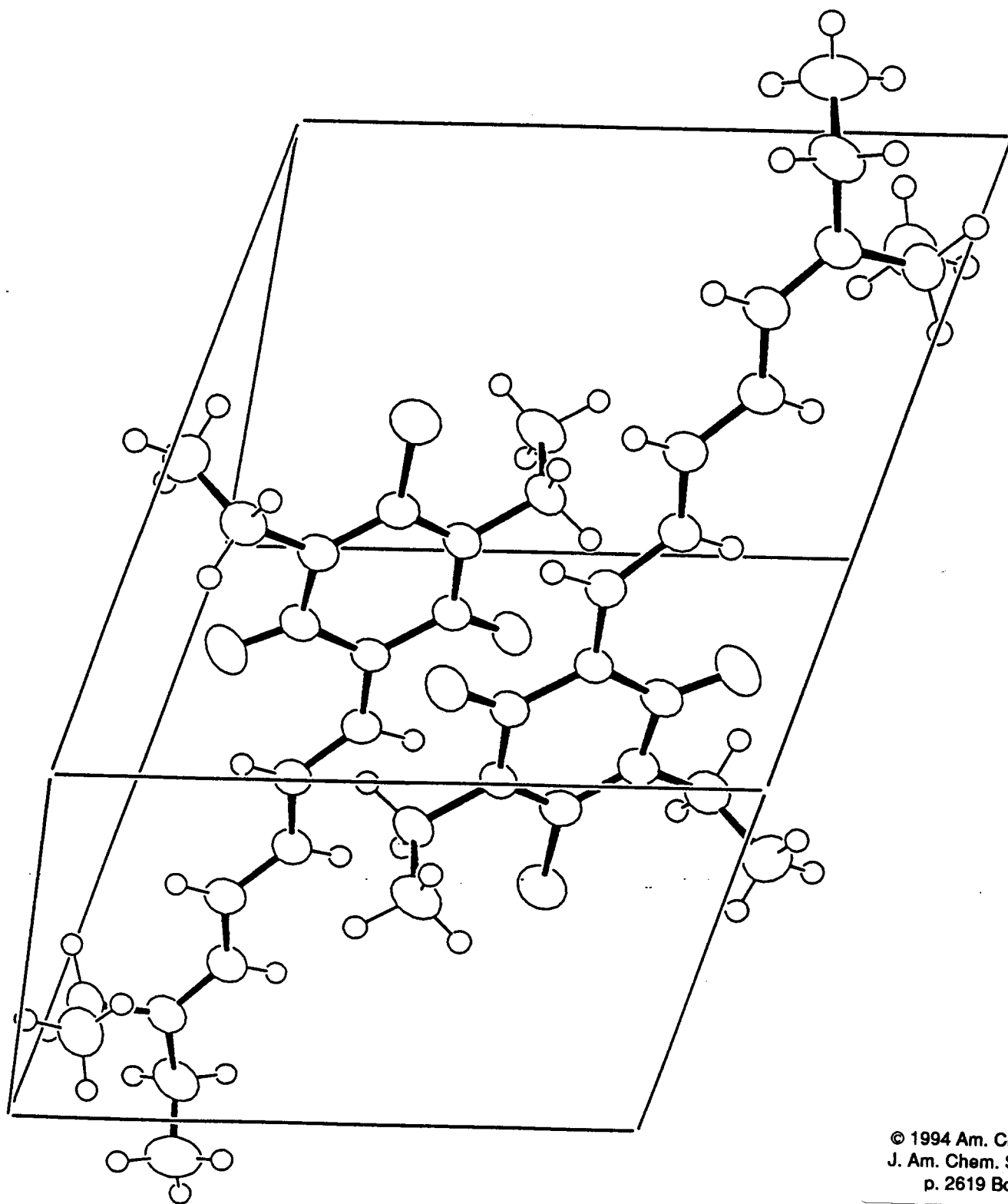
J-2620-m13



J-2620-m14



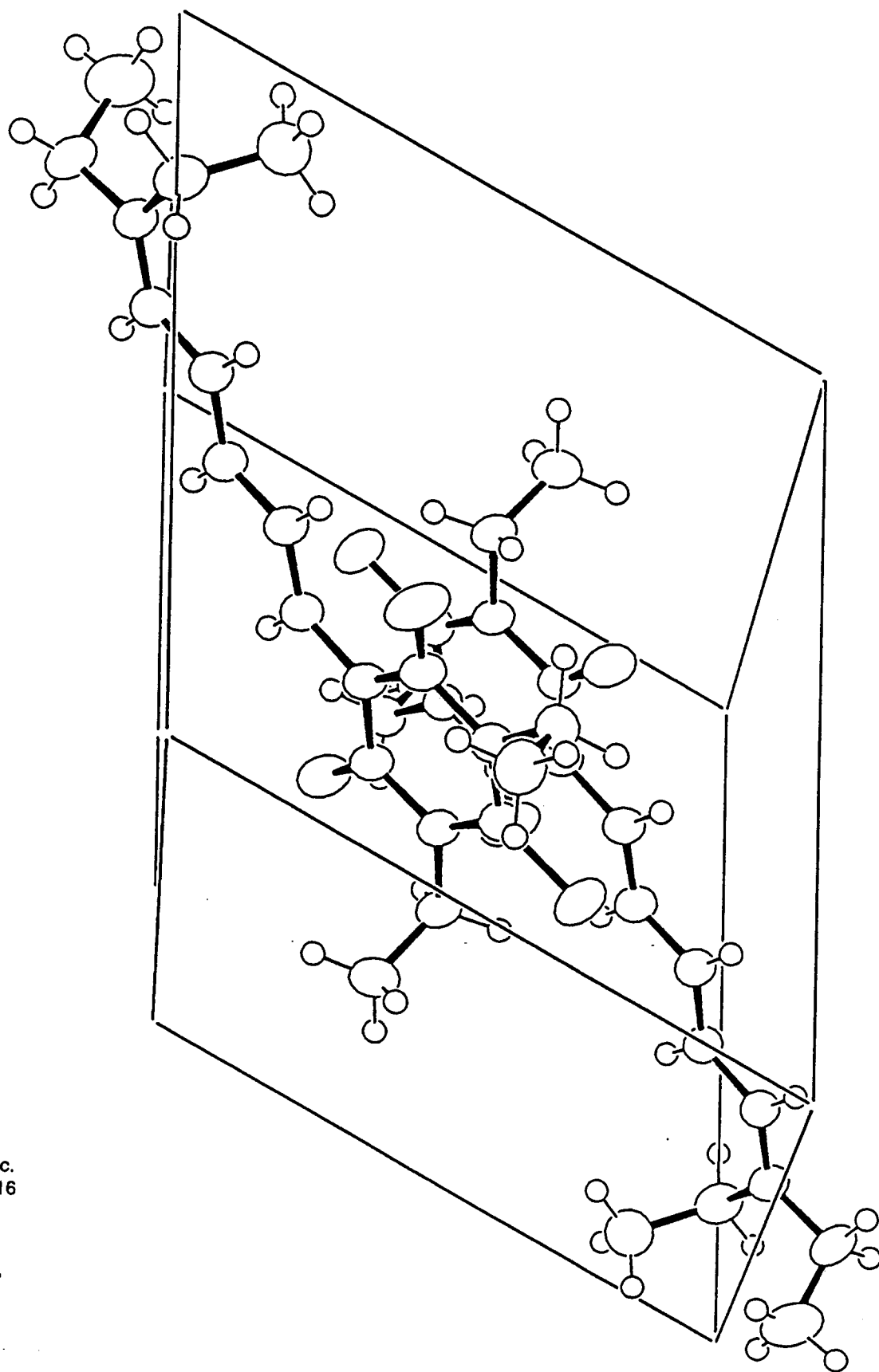
J-2620-m15



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J 2620-m17

Table 1. Final Parameters for  
Diethylamino-II-Thiobarbituric Acid.

$x, y, z$ and $U_{eq}^a \times 10^4$				
Atom	$x$	$y$	$z$	$U_{eq}$ or $B$
N1	1804(1)	2209(1)	-602(1)	426(3)
C1	537(2)	2000(2)	70(2)	494(3)
C2	820(2)	621(2)	1657(2)	626(5)
C3	2260(2)	1608(2)	-1647(2)	581(4)
C4	2763(3)	-172(3)	-894(3)	804(6)
C5	2481(2)	2953(2)	-322(2)	405(3)
C6	2138(2)	3589(2)	618(2)	407(3)
C7	2908(2)	4327(2)	880(2)	408(3)
C8	2597(1)	4920(2)	1863(2)	400(3)
C9	3381(1)	5613(2)	2176(1)	382(3)
C10	3191(1)	6172(1)	3201(1)	361(3)
C11	4195(1)	6775(1)	3394(1)	374(3)
O1	5176(1)	6912(1)	2669(1)	540(2)
N2	4027(1)	7247(1)	4491(1)	370(2)
C12	5032(2)	7956(2)	4597(2)	434(3)
C13	4534(2)	9716(2)	3519(2)	555(4)
C14	2990(1)	7132(1)	5397(1)	381(3)
S	2876(.4)	7680(.5)	6697(.5)	589(1)
N3	2040(1)	6566(1)	5167(1)	385(2)
C15	881(2)	6410(2)	6085(2)	468(3)

J-2620-m18

Table 1. (Cont.)

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>eq</sub> or <i>B</i>
C16	-396(2)	7887(2)	5390(2)	563(4)
C17	2044(1)	6140(2)	4049(2)	415(3)
O2	1068(1)	5782(2)	3876(1)	683(3)
H1A	26(16)	1858(17)	-612(17)	4.7(3)*
H1B	-37(17)	3036(20)	7(17)	5.0(4)*
H2A	-52(22)	593(22)	2024(22)	7.5(5)*
H2B	1261(19)	-372(23)	1636(20)	6.4(5)*
H2C	1366(18)	766(20)	2293(19)	5.7(4)*
H3A	1473(19)	2073(20)	-2414(19)	5.9(4)*
H3B	2980(18)	1968(20)	-2081(18)	5.4(4)*
H4A	3037(22)	-457(25)	-1598(25)	8.3(6)*
H4B	3605(23)	-595(25)	-158(24)	7.8(6)*
H4C	1962(22)	-524(24)	-433(23)	7.8(5)*
H5	3265(15)	3082(16)	-797(15)	3.4(3)*
H6	1366(15)	3483(16)	1141(15)	3.5(3)*
H7	3672(16)	4468(16)	388(16)	3.8(3)*
H8	1794(15)	4818(15)	2362(15)	3.4(3)*
H9	4209(16)	5710(16)	1666(15)	4.0(3)*
H12A	5155(14)	7689(16)	5619(16)	3.9(3)*
H12B	5936(15)	7383(16)	4392(15)	3.8(3)*

J-2620-m19

Table 1. (Cont.)

Atom	x	y	z	$U_{eq}$ or $B$
H13A	5243(17)	10188(19)	3583(17)	5.4(4)*
H13B	4385(16)	9946(18)	2499(19)	4.8(4)*
H13C	3676(18)	10261(19)	3812(18)	5.2(4)*
H15A	693(16)	5504(18)	6148(17)	4.8(4)*
H15B	1254(15)	6150(17)	7048(16)	4.0(3)*
H16A	-1145(20)	7736(21)	6008(20)	6.5(4)*
H16B	-234(18)	8779(22)	5335(20)	6.4(5)*
H16C	-719(19)	8160(22)	4398(22)	7.0(5)*

$$^a U_{eq} = \frac{1}{3} \sum_i \sum_j [U_{ij}(a_i^* a_j^*)(\vec{a}_i \cdot \vec{a}_j)]$$

\* Isotropic displacement parameter,  $B$

J-2620-m20

Table 2. Anisotropic Displacement Parameters for  
Diethylamino-II-Thiobarbituric Acid.

Atom	$U_{11}$	$U_{22}$	$U_{33}$	$U_{12}$	$U_{13}$	$U_{23}$
N1	506(7)	470(7)	409(6)	-233(5)	66(5)	-258(5)
C1	424(8)	571(9)	580(9)	-199(7)	26(7)	-336(8)
C2	676(12)	695(12)	615(11)	-366(10)	202(9)	-336(10)
C3	743(11)	776(12)	560(10)	-430(10)	195(9)	-484(9)
C4	954(17)	832(14)	1013(17)	-407(13)	375(15)	-706(14)
C5	467(8)	421(7)	388(7)	-201(6)	63(6)	-217(6)
C6	470(8)	403(7)	396(7)	-177(6)	67(6)	-220(6)
C7	448(8)	401(7)	416(7)	-158(6)	54(6)	-228(6)
C8	430(7)	391(7)	436(7)	-157(6)	53(6)	-238(6)
C9	398(7)	366(7)	424(7)	-138(6)	67(6)	-226(6)
C10	378(7)	346(6)	418(7)	-140(5)	55(5)	-226(6)
C11	388(7)	361(7)	430(7)	-145(6)	53(6)	-229(6)
O1	514(6)	748(7)	688(7)	-382(5)	258(5)	-513(6)
N2	382(6)	377(6)	422(6)	-157(5)	37(5)	-233(5)
C12	450(8)	475(8)	486(8)	-211(7)	16(6)	-284(7)
C13	690(11)	481(9)	587(10)	-300(8)	49(8)	-270(8)
C14	427(7)	341(7)	375(7)	-116(6)	25(6)	-188(6)
S	657(3)	779(3)	596(3)	-299(2)	139(2)	-511(2)
N3	438(6)	388(6)	400(6)	-186(5)	102(5)	-226(5)
C15	571(9)	504(9)	429(8)	-289(7)	190(7)	-253(7)
C16	534(9)	658(11)	570(10)	-220(8)	201(8)	-364(9)
C17	457(7)	431(7)	482(8)	-211(6)	104(6)	-288(7)
O2	627(7)	1116(10)	937(9)	-594(7)	403(6)	-816(8)

$U_{i,j}$  values have been multiplied by  $10^4$

The form of the displacement factor is:

$$\exp -2\pi^2(U_{11}h^2a^{*2} + U_{22}k^2b^{*2} + U_{33}l^2c^{*2} + 2U_{12}hka^*b^* + 2U_{13}hla^*c^* + 2U_{23}k\ell b^*c^*)$$

J-2620-m21

Table 3. Complete Distances and Angles for  
Diethylamino-II-Thiobarbituric Acid.

Distance(Å)		Distance(Å)	
N1 -C1	1.472(2)	C7 -H7	0.925(16)
N1 -C3	1.469(2)	C8 -H8	0.957(15)
N1 -C5	1.320(2)	C9 -H9	0.984(16)
C1 -C2	1.513(3)	C12 -H12A	0.953(16)
C3 -C4	1.502(4)	C12 -H12B	0.995(16)
C5 -C6	1.384(2)	C13 -H13A	1.021(18)
C6 -C7	1.376(2)	C13 -H13B	0.965(18)
C7 -C8	1.390(2)	C13 -H13C	0.995(19)
C8 -C9	1.376(2)	C15 -H15A	0.985(17)
C9 -C10	1.405(2)	C15 -H15B	0.956(16)
C10 -C11	1.433(2)	C16 -H16A	0.98(2)
C10 -C17	1.429(2)	C16 -H16B	0.96(2)
C11 -O1	1.227(2)	C16 -H16C	0.96(2)
C11 -N2	1.413(2)		
N2 -C12	1.484(2)		
N2 -C14	1.375(2)		
C12 -C13	1.505(3)		
C14 -S	1.671(2)		
C14 -N3	1.372(2)		
N3 -C15	1.483(2)		
N3 -C17	1.416(2)		
C15 -C16	1.503(3)		
C17 -O2	1.224(2)		
C1 -H1A	0.989(17)		
C1 -H1B	1.003(18)		
C2 -H2A	0.96(2)		
C2 -H2B	0.98(2)		
C2 -H2C	0.976(19)		
C3 -H3A	0.97(2)		
C3 -H3B	0.933(19)		
C4 -H4A	0.91(3)		
C4 -H4B	1.00(3)		
C4 -H4C	1.01(2)		
C5 -H5	0.934(16)		
C6 -H6	0.941(16)		

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Table 3. (Cont.)

Angle(°)				Angle(°)			
C3	-N1	-C1	117.4(1)	H2A	-C2	-C1	108.7(14)
C5	-N1	-C1	121.6(1)	H2B	-C2	-C1	108.8(13)
C5	-N1	-C3	121.0(1)	H2C	-C2	-C1	109.6(11)
C2	-C1	-N1	113.9(2)	H2B	-C2	-H2A	106.1(18)
C4	-C3	-N1	113.0(2)	H2C	-C2	-H2A	109.4(18)
C6	-C5	-N1	126.2(1)	H2C	-C2	-H2B	114.1(17)
C7	-C6	-C5	123.1(1)	H3A	-C3	-N1	107.4(12)
C8	-C7	-C6	123.3(1)	H3B	-C3	-N1	106.7(12)
C9	-C8	-C7	124.1(1)	H3A	-C3	-C4	110.1(12)
C10	-C9	-C8	128.3(1)	H3B	-C3	-C4	109.8(12)
C11	-C10	-C9	118.2(1)	H3B	-C3	-H3A	109.8(17)
C17	-C10	-C9	121.9(1)	H4A	-C4	-C3	108.2(16)
C17	-C10	-C11	120.0(1)	H4B	-C4	-C3	107.3(14)
O1	-C11	-C10	124.4(1)	H4C	-C4	-C3	107.9(14)
N2	-C11	-C10	117.1(1)	H4B	-C4	-H4A	107.9(21)
N2	-C11	-O1	118.4(1)	H4C	-C4	-H4A	110.7(21)
C12	-N2	-C11	115.7(1)	H4C	-C4	-H4B	114.7(20)
C14	-N2	-C11	124.4(1)	H5	-C5	-N1	118.5(10)
C14	-N2	-C12	119.9(1)	H5	-C5	-C6	115.3(10)
C13	-C12	-N2	111.3(1)	H6	-C6	-C5	119.3(10)
S	-C14	-N2	121.7(1)	H6	-C6	-C7	117.6(10)
N3	-C14	-N2	116.7(1)	H7	-C7	-C6	120.9(10)
N3	-C14	-S	121.6(1)	H7	-C7	-C8	115.8(10)
C15	-N3	-C14	120.5(1)	H8	-C8	-C7	117.4(9)
C17	-N3	-C14	124.5(1)	H8	-C8	-C9	118.5(9)
C17	-N3	-C15	115.0(1)	H9	-C9	-C8	118.3(10)
C16	-C15	-N3	112.3(1)	H9	-C9	-C10	113.4(9)
N3	-C17	-C10	117.0(1)	H12A	-C12	-N2	106.3(10)
O2	-C17	-C10	125.1(1)	H12B	-C12	-N2	106.4(9)
O2	-C17	-N3	117.9(1)	H12A	-C12	-C13	113.4(10)
H1A	-C1	-N1	105.2(10)	H12B	-C12	-C13	112.2(9)
H1B	-C1	-N1	106.6(10)	H12B	-C12	-H12A	106.9(13)
H1A	-C1	-C2	111.4(10)	H13A	-C13	-C12	110.7(10)
H1B	-C1	-C2	112.0(11)	H13B	-C13	-C12	111.2(11)
H1B	-C1	-H1A	107.3(14)	H13C	-C13	-C12	109.9(11)

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Table 3. (Cont.)

Angle(°)	
H13B -C13 -H13A	108.5(15)
H13C -C13 -H13A	104.8(15)
H13C -C13 -H13B	111.6(15)
H15A -C15 -N3	104.9(10)
H15B -C15 -N3	104.9(10)
H15A -C15 -C16	110.8(10)
H15B -C15 -C16	113.2(10)
H15B -C15 -H15A	110.4(14)
H16A -C16 -C15	110.2(12)
H16B -C16 -C15	111.9(12)
H16C -C16 -C15	111.6(13)
H16B -C16 -H16A	105.8(18)
H16C -C16 -H16A	108.6(18)
H16C -C16 -H16B	108.6(18)

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**Table 4. Intermolecular Distances Less than 3.5 Å for  
Diethylamino-II-Thiobarbituric Acid.**

Distance(Å)		Distance(Å)	
C1 -C8	3.481(2)	C8 -H1A	3.191(17)
C3 -O1	3.344(2)	C8 -H1B	2.822(18)
C5 -O1	3.307(2)	C8 -H16A	3.32(2)
C9 -C12	3.452(2)	C8 -H12A	3.000(16)
C10 -N2	3.477(2)	C8 -H12B	3.479(16)
C11 -C11	3.440(2)	C9 -H4C	3.37(2)
C11 -N2	3.441(2)	C9 -H1A	3.474(17)
O1 -C14	3.458(2)	C9 -H12A	2.903(16)
O1 -N3	3.473(2)	C9 -H12B	3.274(16)
C15 -O2	3.465(2)	C10 -H2B	3.04(2)
O2 -O2	3.312(2)	C10 -H12A	3.345(16)
C1 -H8	3.189(15)	C10 -H12B	3.094(16)
N1 -H13B	3.493(18)	C11 -H2B	3.24(2)
N1 -H15A	3.286(17)	C11 -H4B	3.25(3)
N1 -H15B	3.422(16)	O1 -H4B	2.85(3)
C1 -H4C	3.31(2)	O1 -H3B	2.437(19)
C2 -H16C	3.48(2)	O1 -H5	2.440(16)
C2 -H4C	3.23(2)	N2 -H2B	3.46(2)
C2 -H16B	3.43(2)	N2 -H2C	3.437(19)
C3 -H15A	3.327(17)	C12 -H3B	3.200(19)
C3 -H16C	3.18(2)	C12 -H13A	3.208(18)
C3 -H12B	3.222(16)	C13 -H2C	3.180(19)
C4 -H2A	3.31(2)	C13 -H3B	3.296(19)
C4 -H13A	3.360(18)	C13 -H4A	3.16(3)
C4 -H13B	3.308(18)	C13 -H12A	3.303(16)
C5 -H13B	3.110(18)	C13 -H13A	3.055(18)
C5 -H15A	3.419(17)	C13 -H13C	3.398(19)
C5 -H15B	2.948(16)	C14 -H2C	3.432(19)
C6 -H13B	3.285(18)	S -H4A	3.21(3)
C6 -H13C	3.323(19)	S -H2A	3.43(2)
C6 -H15B	3.260(16)	S -H9	3.441(16)
C6 -H1B	3.181(18)	S -H16B	3.41(2)
C6 -H16A	3.30(2)	S -H13A	3.283(18)
C7 -H1B	3.074(18)	N3 -H2B	3.36(2)
C7 -H12A	3.462(16)	C15 -H8	3.405(15)



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Table 4. (Cont.)

Distance(Å)		Distance(Å)	
C16 -H2A	3.32(2)	H3A -H12B	3.30(3)
C16 -H3A	3.29(2)	H3B -H15A	3.23(3)
C16 -H2C	3.262(19)	H3B -H16C	3.49(3)
C16 -H6	3.377(16)	H3B -H12B	2.45(3)
C16 -H8	3.349(15)	H3B -H13A	3.31(3)
C16 -H16B	3.42(2)	H3B -H13B	2.92(3)
C17 -H2B	3.06(2)	H4A -H16C	3.15(3)
C17 -H1A	3.474(17)	H4A -H13A	2.60(3)
C17 -H12B	3.117(16)	H4A -H13B	2.89(3)
O2 -H1A	3.018(17)	H4B -H9	3.16(3)
O2 -H3A	2.70(2)	H4B -H13B	3.23(3)
O2 -H15A	2.579(17)	H4B -H5	3.20(3)
O2 -H12B	3.394(16)	H4B -H13B	3.06(3)
H1A -H1A	3.41(2)	H4C -H9	3.38(3)
H1A -H2A	3.46(3)	H5 -H13B	3.25(2)
H1A -H2B	2.87(3)	H5 -H15A	3.46(2)
H1A -H4C	2.72(3)	H5 -H15B	2.91(2)
H1A -H8	2.92(2)	H5 -H9	3.15(2)
H1B -H6	2.98(2)	H6 -H13C	3.28(2)
H1B -H8	2.58(2)	H6 -H6	3.30(2)
H2A -H16B	3.02(3)	H6 -H15B	3.22(2)
H2A -H16C	2.78(3)	H6 -H16A	2.60(3)
H2A -H3A	3.37(3)	H7 -H7	3.19(2)
H2A -H4A	3.18(3)	H7 -H9	2.98(2)
H2A -H4C	2.66(3)	H8 -H15A	2.94(2)
H2A -H16B	3.13(3)	H8 -H16A	2.63(3)
H2B -H16C	3.43(3)	H8 -H12A	3.30(2)
H2B -H4C	3.26(3)	H8 -H12B	3.42(2)
H2C -H13B	2.92(3)	H9 -H12A	3.11(2)
H2C -H13C	2.67(3)	H12A -H13A	2.59(2)
H2C -H16B	3.49(3)	H12A -H13C	3.03(3)
H2C -H16A	2.80(3)	H13A -H13A	2.79(3)
H2C -H16B	2.83(3)	H13A -H13C	2.72(3)
H3A -H15A	2.91(3)	H13C -H16A	2.79(3)
H3A -H16C	2.37(3)	H13C -H13C	3.48(3)

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Table 4. (Cont.)

Distance(Å)

H16B -H16B 2.47(3)

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**Table 5. Observed and Calculated Structure Factors for  
Diethylamino-II-Thiobarbituric Acid**

The columns contain, in order,  $\ell$ ,  $10F_{obs}$ ,  $10F_{calc}$  and  $10\sigma F_{obs}$ . A minus sign preceding  $F_{obs}$  indicates that  $F_{obs}^2$  is negative.

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Diethylamino-II-Thiobarbituric Acid

Page 2

3	14	4	2	4	48	49	1	5	-6	4	2	9	-6	1	3
4	16	14	1	5	62	62	1	6	9	9	2	10	51	49	1
5	100	98	1	6	193	195	1	7	128	127	1	11	23	24	1
6	0	2	7	7	33	35	1	8	-5	3	3	12	32	29	1
7	32	31	1	8	14	3	2	9	16	13	1				
8	36	36	1	9	26	27	1	10	78	77	1				
9	12	15	2	10	24	22	1	11	-9	3	2		-5	7	1
10	36	34	1	11	-10	4	2								
	-7	7	1		-6	5	1		-5	2	1				
1	-5	6	4	1	41	45	1	1	16	9	1	1	38	38	1
2	56	56	1	2	44	45	1	2	4	4	2	2	51	49	1
3	15	4	2	3	9	2	2	3	87	81	1	3	12	11	2
4	43	38	1	4	15	8	1	4	127	128	1	4	50	45	1
5	69	69	1	5	67	66	1	5	118	121	1	5	31	25	1
6	19	20	1	6	129	129	1	6	140	142	1	6	60	58	1
7	46	45	1	7	19	20	1	7	158	162	1	7	30	34	1
8	-8	9	3	8	99	95	1	8	71	71	1	8	41	42	1
9	21	20	1	9	33	34	1	9	60	59	1	9	24	26	1
	-7	8	1	10	12	12	2	10	20	22	1	10	7	7	3
				11	54	51	1	11	23	22	1	11	27	20	1
								12	17	17	2				
2	49	48	1		-6	6	1		-5	3	1				
3	42	40	1									1	19	16	1
4	45	44	1	1	11	14	2	1	139	138	1	2	56	53	1
5	32	31	1	2	22	24	1	2	84	81	1	3	5	3	4
6	0	0	8	3	128	125	1	3	50	51	1	4	74	79	1
7	8	8	3	4	47	48	1	4	119	116	0	5	108	109	1
8	7	2	4	5	16	3	1	5	89	88	1	6	75	77	1
	-6	1	1	6	26	20	1	6	74	75	1	7	12	17	2
1	39	35	0	7	12	10	2	7	221	223	2	8	7	15	3
2	115	108	1	8	17	17	1	8	21	22	1	9	31	26	1
3	163	168	1	9	17	16	1	9	49	48	1	10	65	57	1
4	33	33	1	10	56	53	1	10	33	33	1	11	33	32	1
5	32	35	1	11	24	23	1	11	54	53	1		-5	9	1
6	84	86	1		-6	7	1	12	34	33	1	1	10	7	3
7	75	72	1									2	31	30	1
8	90	91	1	1	8	7	3		-5	4	1	3	5	3	4
9	11	12	2	2	-6	2	3	1	73	75	1	4	7	7	3
10	55	55	1	3	46	44	1	2	63	64	1	5	15	18	2
11	12	15	2	4	98	97	1	3	59	57	1	6	8	22	3
	-6	2	1	5	137	137	1	4	168	171	1	7	12	13	2
1	207	201	1	6	51	52	1	5	16	11	1	8	20	21	1
2	174	173	1	7	14	13	2	6	95	94	1	9	14	8	2
3	60	63	1	8	35	36	1	7	96	98	1	10	16	10	2
4	5	3	3	9	-10	0	2	8	14	14	1		-5	10	1
5	23	21	1	10	14	9	2	9	53	48	1	4	31	29	1
6	94	97	1		-6	8	1	10	35	32	1	5	-6	5	4
7	20	18	1	1	10	13	3		33	33	1	6	18	14	2
8	70	71	1	2	6	4	4		34	34	1	7	33	31	1
9	76	75	1	3	12	8	2					8	18	19	2
10	-5	6	4	4	34	29	1	1	61	59	1		-4	1	1
11	-6	9	3	5	54	53	1	2	16	19	1	1	283	273	2
	-6	3	1	6	40	42	1	3	43	43	1	2	340	341	2
1	84	78	1	7	27	27	1	4	169	170	1	3	146	142	1
2	25	25	1	8	13	8	2	5	206	207	1	4	133	136	1
3	18	22	1	9	58	54	1	6	160	157	1	5	136	132	1
4	39	38	1	10	24	23	1	7	138	137	1	6	22	25	1
5	92	91	1		-6	9	1	8	-7	9	2	7	27	27	1
6	16	11	1					9	33	32	1	8	282	283	2
7	44	43	1	3	30	30	1	10	7	11	3	9	70	68	1
8	92	91	1	4	35	33	1	11	10	13	2	10	14	10	2
9	120	120	1	5	42	39	1	12	35	32	1	11	35	32	1
10	6	12	3	6	8	1	3					12	-8	2	3
11	16	15	2	7	-7	6	3		-5	6	1				
	-6	4	1	8	35	33	1	1	18	17	1		-4	2	1
1	43	44	1		-5	1	1	2	14	10	1	1	245	234	2
2	16	19	1	1	239	232	2	3	41	39	1	2	137	129	1
3	54	56	1	2	202	188	1	4	30	33	1	3	46	40	0
				3	126	115	1	5	146	146	1	4	137	132	1
				4	133	135	1	6	212	214	2	5	61	62	0
								7	54	54	1	6	92	87	1
								8	-10	5	2	7	119	117	1

[illegible]

1	6	0	1					7	249	246	2	6	93	98	1
2	119	120	1	1	34	34	1	8	52	55	1	7	4	6	3
3	52	51	0	2	68	65	1	9	32	30	1	8	126	131	1
4	-3	3	2	3	80	79	1	10	25	22	1	9	-10	10	2
5	285	282	2	4	26	27	1	11	24	29	1	10	57	57	1
6	232	225	2	5	65	67	1	12	19	19	2	11	56	56	1
7	118	118	1	6	34	37	1					12	-11	3	2
8	91	93	1	7	60	59	1	-1	3	1		13	18	12	2
9	43	44	1	8	8	10	2				0				
10	91	93	1	9	24	23	1	1	22	25	0	-1	8	1	
11	16	17	2	10	-6	1	3	2	289	285	2				
12	44	45	1	11	31	29	1	3	98	93	1	1	45	46	1
				12	44	39	1	4	99	98	1	2	62	62	1
-2	4	1						5	170	176	1	3	59	57	1
				-2	9	1		6	60	65	0	4	75	74	1
1	12	13	1					7	34	37	0	5	10	14	2
2	28	30	0	1	6	8	3	8	61	61	1	6	66	67	1
3	195	203	1	2	20	16	1	9	64	63	1	7	15	16	1
4	81	79	1	3	10	8	2	10	-11	0	1	8	121	129	1
5	44	50	0	4	38	35	1	11	63	60	1	9	41	42	1
6	194	203	1	5	10	0	2	12	28	25	1	10	67	69	1
7	126	125	1	6	2	6	5	13	6	0	4	11	17	21	1
8	8	1	2	7	125	129	1					12	30	31	1
9	28	27	1	8	40	39	1	-1	4	1		13	41	43	1
10	38	40	1	9	43	43	1								
11	45	43	1	10	47	47	1	1	164	166	1	-1	9	1	
12	38	37	1	11	4	6	4	2	41	38	0				
13	42	36	1	12	7	12	3	3	9	10	1	1	29	31	1
								4	478	462	4	2	80	80	1
-2	5	1		-2	10	1		5	161	164	1	3	91	91	1
1	17	16	1	1	33	30	1	6	47	41	0	4	3	4	4
2	28	28	0	2	44	44	1	7	8	13	1	5	98	96	1
3	70	63	0	3	42	39	1	8	18	17	1	6	70	73	1
4	166	164	1	4	25	25	1	9	71	73	1	7	2	4	5
5	53	54	0	5	57	57	1	10	40	40	1	8	5	4	3
6	85	83	1	6	19	21	1	11	23	23	1	9	7	13	3
7	51	48	1	7	60	64	1	12	26	26	1	10	43	40	1
8	43	42	1	8	9	16	3	13	53	48	1	11	42	40	1
9	8	2	2	9	30	30	1					12	13	15	2
10	46	46	1	10	-2	11	6	-1	5	1		-1	10	1	
11	8	8	3	11	12	13	2	1	40	41	0				
12	54	48	1					2	22	22	0	1	9	9	3
13	73	70	1	-2	11	1		3	110	103	1	2	11	13	2
								4	232	224	2	3	11	6	2
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Page 5															
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## Diethylamino-II-Thiobarbituric Acid

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Supplementary material,  
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3	8	5	3	2	0	1		2	3	1		2	6	1	

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Thylamino-II-Thioarbituric ACID																
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13	44	43	1		1	8	8	2						8		

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-3	37	35	1	-2	8	4	3	0	32	28	1	2	43	43	1
-2	13	9	2	-1	12	9	2	1	18	16	2	3	-9	6	3
-1	12	5	2	1	39	41	1	2	26	26	1	4	41	36	1
0	57	55	1	2	49	48	1	3	13	14	2				
1	19	17	2	2	18	14	2	4	9	13	3	13	7	1	
2	7	4	3	3	-5	5	4								
3	9	7	3					13	6	1		0	-6	5	4
				13	5	1						1	-4	1	5
13	4	1		-2	14	13	2	-1	-2	6	6	2	49	45	1
				-1	19	18	2	0	6	3	4	3	24	24	1
								1	58	53	1	4	17	21	2

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